Pyrrolopyrimidine Nucleosides. 1. The Synthesis of 4-Substituted $7-(\beta-D-Ribofuranosyl)$ pyrrolo[2,3-d]pyrimidines from Tubercidin'

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The synthesis of 4-chloro-7-(β -p-ribofuranosyl)pyrrolo[2,3-d]pyrimidine (V1a) has been accomplished by treatment of 7-(2',3',5'-tri-O-acetyl- β -p-ribofuranosyl)pyrrolo[2,3-d]-4-pyrimidine (V) with POCl₅. Nucleophilic substitution of the 4-chloro group resulted in the synthesis of new analogs of tubercidin. Direct methylation of tubercidin (I) alforded 4-amino-3-methyl-7-(β -p-ribofuranosyl)pyrrolo[2,3-d]pyrimidine (XI) when treated with NatOll. The synthesis of XI was also accomplished directly from VIa with methylamine.

Isolation² of the nucleoside antibiotic tubereidin from *Streptomyces tubereidicus* provided the first naturally occurring derivative of the pyrrolo[2,3-d]pyrimidine ring system. The structure of tubereidin was later established³ unequivocally as 4-anino-7-(β-pribofuranosyl)pyrrolo[2,3-d]pyrimidine (7-deazadenosine, I). Tubereidin has subsequently demonstrated⁴



growth inhibition of certain experimental animal tumors. The *in viva* inhibition⁵ of mouse fibroblast multiplication and definite inhibition⁶ of several human tumors in vitro has also been noted. Tubercidin inhibits de nava purine biosynthesis by preventing the synthesis of 5-phosphoribosylpyrophosphate (PRPP).⁷ It has been shown⁸ that the binding of certain animoacyl-RNA's to ribosomes was significantly stimulated when tubercidin was incorporated into the aminoacyl-s-RNA. The isolation and characterization of other derivatives of the pyrrolo [2,3-d] pyrimidine ring system, toyocamycin^a (II) and sangivamycin¹⁰ (III), from naturally occurring sources has created considerable interest in the chemical synthesis of these compounds and their derivatives. It has been

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Acetvlation of 7-(β-n-ribofuranosyl)pyrrolo[2,3-d]-4pyrimidone (IV, 7-deazainosine)¹² with a mixture of acetic anhydride and pyridine furnished a good yield of 7-(2'.3'.5'-tri-O-acetyl-β-p-ribofuranosyl)pyrrolo[2,-3-d]-4-pyrimidone (V). Chlorination with POCl₃ and redistilled N.N-diethylaniline has been previously reported¹³ to afford 6-chloro-9-(2',3',5'-tri-O-acetyl-3-D-ribofuranosyl)purine from 9-(2',3',5'-tri-O-acetyl-3-n-ribofuranosyl)-5-purinone. However, these reaction conditions with V produced a highly colored solution from which none of the desired Vlb could be isolated. It was subsequently discovered that by simply excluding N₂N-diethylaniline from the reaction mixture a 70% yield of the desired product, 4-chloro-7-(2',3',5'-tri-O-acetyl-3-p-ribofuranosyl) pyrrolo[2,3d]pyrinidine (VIb), was readily obtained (Chart I). It is of interest that an attempted conversion of 4-methylthio-7-(2',3'-di-O-acetyl-5'-O-trityl-3-p-ribofuranosyl)pyrrolo[2,3-d]pyrimidine to the corresponding 4-chloro derivative using a previously reported procedure¹⁴ was also unsuccessful, presumably due to oxidation of the pyrrole ring by excess chlorine. Removal of the blocking groups on the carbohydrate moiety with methanolic ammonia furnished the versatile compound. 4-chloro-7-(β-n-ribofuranosyl)pyrrolo[2,3d]pyrimidine (VIa). The recent finding¹⁵ that 6chloro-9-(β -n-ribofuranosyl)purine is dechlorinated to inosine by heart, red cell, and intestinal mncosa adenosine deaminases would indicate that the chemotherapeutic activity of v-chloro-9-(g-n-ribofuranosyl)purine

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might be increased if this reaction could be inhibited or eliminated. It is of interest that tubercidin demonstrated a complete resistance to these three deaminases under the same conditions and, therefore, by analogy VIa might well resist enzymatic dechlorination by adenosine deaminase.

Treatment of VIa with aqueous thiourea produced $7 - (\beta - p - ribofuranosyl) pyrrolo [2, 3-d] pyrimidine - 4-thiol$ (VII) which was identical in all respects with the same compound previously prepared^{12e} from the thiation of V with P₂S₅ followed by removal of the blocking groups on the carbohydrate moiety. 4-Benzylthio-7- $(\beta$ -Dribofuranosyl)pyrrolo[2,3-d]pyrimidine (IXb) was produced when VII was treated with benzyl chloride in dilute aqueous ammonia. It is of considerable interest that 6-methylthiopurine ribonucleoside has been postulated¹⁶ to act as a substrate for adenosine kinase which converts it to the nucleotide form in mouse tissues, Ehrlich aseites carcinoma cells in vivo, and tumor cells in vitro. 6-Methylthiopurine ribonucleoside has also demonstrated¹⁷ significant in vivo activity against leukemia L1210 which had become resistant to 6-mercaptopurine therapy. This activity has prompted an investigation to determine if a correlation between the ring nitrogens and cytotoxicity might exist. While it has been determined that 3-deaza-6methylthiopurine ribonucleoside¹⁸ possesses far greater activity than does 1-deaza-6-methylthiopurine ribonucleoside,¹⁹ neither compound compares favorably with the parent compound, 6-methylthiopurine ribonucleoside. The only remaining deaza nucleoside derivative is the compound, 4-methylthio-7-(β -pribofuranosyl)pyrrolo[2,3-d]pyrimidine (IXa) which was prepared in our laboratory by alkylation of VII with methyliodide.

Dehalogenation of VIa with hydrogen and Pd–C catalyst has resulted in the preparation of 7-(β -D-ribofuranosyl)pyrrolo[2,3-d]pyrimidine (VIIId, 7-de-azanebularine). This compound is of interest since the active constituent from Agaricus nebularis Batch has been isolated²⁰ and later characterized²¹ as the nucleoside antibiotic 9-(β -D-ribofuranosyl)purine (nebularine). Nebularine has also been isolated²² from Streptomyces yokosukaensis and has demonstrated²³ some activity in tissue cultures. In fact it has been recently suggested²⁴ that 7-deazanebularine (VIId) should be synthesized and its biochemical behavior compared with that of nebularine.

Nucleophilic displacement of the 4-chloro group of VIa in refluxing sodium methoxide afforded 4-methoxy- $7-(\beta-p-ribofuranosyl)$ pyrrolo [2,3-d] pyrimidine (VIIIa). Treatment of VIa with dimethylamine and piperidine under anhydrous conditions produced 4-dimethylanino- $7-(\beta-p-ribofuranosyl)$ pyrrolo [2,3-d]pyrimidine (VIIIb) and 4-(1-piperidvl)-7-(β -D-ribofuranosvl)pyrrolo[2,3-d]pyrimidine (VIIIc), respectively. It is of interest that the corresponding purine ribonucleoside, 6-(1-piperidyl)-9-(β -p-ribofuranosyl)purine possesses²⁵ significant antitumor activity against Walker 755 and 756 carcinoma and inhibits the growth of Vicia faba. Treatment of VIa with methanolic ammonia at 150° resulted in a displacement of the chloro group to produce 4-amino-7-(β -D-ribofuranosyl)pyrrolo [2,3-d]pyrimidine (I, tubereidin).

Actually this provides the first total chemical synthesis of tubercidin since 7- $(\beta$ -p-ribofuranosyl)pyrrolo-[2,3-d]-4-pyrimidone has been previously prepared^{12b} by chemical synthesis.

Methylation of tubercidin (I) with methyl iodide in dimethylacetamide (Chart II) furnished a monomethyl derivative as the iodide salt which was assumed to be 4-anino-3-methyl-7- $(\beta$ -p-ribofuranosyl)pyrrolo[2,3-d]pyrimidine (X). The pmr spectrum indicated that alkylation of a ring nitrogen had occurred since the absorption peak observed for the methyl group was a singlet and it has been previously shown²⁶ that under similar reaction conditions adenosine undergoes alkylation on the ring nitrogen adjacent to the exocyclic

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amino group to afford 1-methyladenosine. Treatment of X with strong base produced another compound with a different melting point, pmr (methyl group occurs as a doublet), and ultraviolet spectra. The structure of this compound was unequivocally established as 4methylamino-7-(β -p-ribofuranosyl)pyrrolo[2,3-d]pyrimidine (XI) when the product obtained from the treatment of VIa with anhydrous methylamine was identical in all respects. This firmly established the structure of the unrearranged compound as 4-amino-3-methyl-7-(β -p-ribofuranosyl)pyrrolo[2,3-d]pyrimidine (X).

All of the compounds prepared have purine nucleoside counterparts which possess a wide variety of biological and chemotherapeutic activity. One of the most serions disadvantages in the utilization of purine nucleosides in chemotherapy is the ease with which the glycosidic linkage is cleaved by the enzyme nucleoside phosphorylase. Tubercidin has been found¹²c to be completely resistant toward glycosidic cleavage by this enzyme and it is tempting to postulate that N-7 of the purine ring is therefore an essential binding site for this enzymatic action. Thus all the 7-deazapurine nucleosides prepared in this investigation should possess the advantage of being stable *in vitro* or *in viva* to nucleoside phosphorylase.

Antitumor Evaluation.²⁷—The results obtained from the antitumor testing of compounds prepared in this investigation are shown in Table I and evaluation of the activity is in accordance with the criteria²⁸ of the Cancer Chemotherapy National Service Center. From the testing data available (Table I) at the present time it is evident that this series possesses very little potential as inhibitory agents of lymphoid leukenia L1210. Although VIIIb possessed a T/C of 120%, this fails to pass stage 1 of the sequential screen.

Preliminary testing results against Walker carcino-

(27) Testing was performed under the anspices of the Cancer Chemotherapy National Service Center.

sarcoma 256 (intramuscular), however, have been rather encouraging. These tests were performed on randomly bred albino rats and the compounds were administered intraperitoneally using saline as the vehicle. Allcompounds tested thus far have exhibited some inhibition: in fact. IV has demonstrated sufficient activity to pass stage 1, 2, and 3 of the sequential screen of step 1 and the three successive tests at sequential dose of step 11 confirmation, and $7 - (\beta - p - ribofuranosyl) pyr$ rolo[2,3-d]-4-pyrimidoue (IV) can now be considered a confirmed active compound in this test system. Additional testing results of this series of compounds are at the present time incomplete. However, it does appear that derivatives of the pyrrolo [2,3-d] pyrimidine nucleosides may indeed possess important biological activity.

Experimental Section

Melting points were taken on a Thomas–Hoover melting point apparatus and are uncorrected. Chromatograms were developed using Whatman No. 1 chromatography paper in the following solvent systems: A, $5^{e_{t}}$ aqueous NH₄Cl: B_t ethanol-water (7;3|v/v); C, $5^{e_{t}}$ aqueous NaHCO₃; D, 2-propanol-concentrated NH₄OH–water (4;3;3, v/v); all systems descending. The ultraviolet spectra are given for some of the compounds in Table II.

7-(2',3',5'-Tri-O-acetyl- β -D-ribofuranosyl)pyrrolo[2,3-d]-4pyrimidone (V),—Ten grams of IV was added to a solution of 100 ml of pyridine and 50 ml of acetic anhydride. This mixture was allowed to stand at 5° for 24 hr with occasional shaking until all the solid was in solution. The solution was then evaporated to a thick symp *in vacuo*, and 400 ml of anhydrons ethanol was added. This solution was refuxed for 1 br, the ethanol was removed *in vacuo*, and the symp) was dissolved in 400 ml of methylene chloride. The CH₂Cl₂ solution was washed once with 1 N HCl (300 ml), twice with 300-ml portions of saturated NaHCO₅, and once with 300 ml of H₂O. Another 300 ml of H₂O was added and the mixture was allowed to stir vigorously overnight. The layers were separated and the CH₂Cl₂ portion was dried (MgSO₄). The CH₂Cl₂ was removed *in vacuo* and the residue was dried to a form *in vacuo*; yield 12 g.

Anal. Caled for U₃₇H₁₉N₂O₆; C, 51.9; H, 4.87; N, 10.68, Found: C, 51.87; H, 4.70; N, 10.48.

4-Chloro-7-(2',3',5'-tri-\partial-acetyl-\beta-b-ribofuranosyl)pyrrolo-[2,3-d]pyrimidine (VIb), ..., 7-(2',3',5'-Tri-\partial-acetyl-\beta-b-ribofuranosyl)pyrrolo[2,3-d]-4-pyrimidone (V, 3.0 g) was dissolved in POCl₃ (15 ml) and the solution was heated on a hot water bath at 70-75° for 30 min. The amber solution was then poured over crushed ice (150 ml) with vigorous stirring. The mixture was stirred until the excess POCl₃ was decomposed and additional ice was added if necessary to keep the temperature between 0-5°. The cold mixture was then extracted with two 7\partial-ml portions of CH₂Cl₂ and the combined extracts were allowed to stir overnight with 150 ml of water. The CH₂Cl₂ layer was then separated, dried (MgSO₄), and evaporated *in racia* **to yield 2.0-2.2 g (7\partial T_{i}) of product as an oil which was used directly in the next step.**

4-Chloro-7-(β -()-ribofuranosyl)pyrrolo[2,3-d]pyrimidine (VIa).

4-Chloro-7-(2',3',5'-tri-0-ace(yl-β-o-ribofuranosyl)pyrrol(2,3-d)pyrimidine (VIb, 3.6 g, 0.88 mole) was dissolved in 100 ml of methanolic NH₃ (saturated at -10°) and the solution was allowed to stand at 5° for 24 hr. The solution was then evaporated *in vacuo* at 50-00° to yield a thick symp. The symp was dissolved in 35-50 ml of boiling water and the solution was allowed to cool to room temperature. A colorless solid, which crystallized from the solution, was filtered from the mixture, washed with water, and dried. A yield of 1.8 g of colurless product, mp 161-463°, was obtained. An additional 350 mg of product, mp 159-462°, was obtained by partial evaporation of the filtrate and a total yield of 2.15 g (86°;) was obtained. This product was essentially pure but can be recrystallized from water or a ligroin (bp 90-420°)-acetone mixture to give an analytically pare product, mp 161-463°, $|\alpha|^{24}$ p -63.2° (c 4, 1):1 ethanol-HgO).

⁽²⁸⁾ The requirements for passing different stages of the sequential screen where the value of T^+C in a single test at a matroxic dose must be as follows: hymphoid backenia L1240, stage 1 $T^+C \ge 125\%$, and stage 2 $T^+C \ge 156\%$: Walker careinosarcoma 256 (intra)misedar), stage 1 $T^+C \ge 0.53$, stage 2 $T^+C \le 0.19$, and stage 3 $T^+C \le 0.07$.

TABLE I TESTING DATA OF SOME 4-SUBSTITUTED 7-(\$-D-R1BOFURANOSY1.)PYRROLO[2,3-d]PYRIMIDINES



				Animal				
<i>(</i>)		Dose,	61	wt diff	Survival days	117	Stage	Test
Compa	R	ing/kg	Lymph	(1 - 0)	210	%	inilex	status
1.11	au	5.00	e /e	o e	10 7 (0 0	110	110	0
\ 11 \-TTT	Sn	000 400	0/0	-2.0	10.7/9.4	110	110	<u> </u>
VIIIa	$0CH^3$	400	0/0					1
		200	0/4		$\omega = \omega$	101		22
		100	3/4	-0.0	8.0/8.4	101		22
		00 07	4/4	-3.7	8.0/8.4	101		22
VIa	01	20	3/4 = /6	-0.0	(.3/8.4	80		22
	CI	400	5/0	-4.2	0.2/8.2	110		1
		200	4/4	-3.8	9.3/8.4	110		22
		100	4/4	-2.8	9.5/8.4	110		22
		00 97	4/4	-0.8	8.4/8.4	104		22
T 3.71	COLOID	25	4/4	0.2	9.3/8.4	110		22
IXb	SCH ₂ C ₆ H ₅	400	1/6	-3.2	6.0/8.2	110		1
		200	4/4	-2.2	9.8/8.4	110		22
		100	4/4	-0.4	8.3/8.4	98		22
		50	4/4	-1.6	9.0/8.4	107		22
T 1 /	011	25	4/4	-2.2	9.3/8.4	110		22
IV	ОН	250	4/6	-4.8	6.0/8.9			3
		100	0/6					3
		ə00	0/6	2.0		100		1
	Non	12.5	6/6	-2.0	9.5/9.3	102		
VIIIc	NC_5H_{10}	400	6/6	-0.2	8.5/8.9	95		2
		400	4/4	0.4	8.3/8.9	93		22
		200	4/4	-0.1	8.8/8.9	98		22
		100	4/4	-0.2	8.5/8.9	95		22
V/T	NUMBER	- 0 0	4/4	0.5	8.3/8.9	93		22
XI	$\rm NHCH_3$	400	1/4	-6.0	9.0/8.9			22
		200	0/4	F ()				22
		100	1/4	-5.9	11.0/8.9			22
		50	$\frac{4}{4}$	-1.8	9.8/8.9			22
	NUCLE	40	0/6					1
VIIIb	$N(CH_3)_2$	500	0/6					1
		125	0/6	2.0				3
		30	1/6	-6.9	7.0/8.9	10.		3
		7.5	6/6	-1.7	11.2/9.3	120		4
	Walker Carcinosarcoma 256 (intramuscular)							
					Tumor wt (g) T/C			
VIIIb	N(CH ₂) ₂	7.5	5/5	-19	5.2/9.9	52	52	11
		7.5	6/6	-12	8.0/9.4	85	44	6
VII	SH	500	$\frac{2}{6}$	-24()	1 0/7 2			1
. ==	~==	250	5/6	-12.0	1.8/5.0	36	36	13
IV	OH	65	7/7	-2.0	4.4/9.0	48	48	11
	2	12.5	5/7	-26.0	0.9/9.0	10	10	11
		12.5	6/6	-22.0	1.8/5.1	35	3	15
		12.5	$\frac{4}{6}$	-25.0	,			-0
		12.5	$\frac{3}{6}$	-23.0	2.1/5.5	38	38	20
		10.0	4/6	-21.0	2.9/9.3	31		20.4
		10.0	6/6	-19.0	1.6/5.0	32		200

Anal. Calcd for $C_{11}H_{12}ClN_3O_4;\ C,\ 46.3;\ H,\ 4.23;\ N,\ 14.7;\ Cl,\ 12.4.$ Found: C, 46.1; H, 4.07; N, 14.8; Cl, 12.4.

 $\textbf{4-Amino-7-} (\beta \textbf{-} \textbf{D} \textbf{-} \textbf{ribofuranosyl}) \textbf{pyrrolo} [\textbf{2,3-}d] \textbf{pyrimidine} \quad (\textbf{Tu-}$ bercidin, I) —4-Chloro-7-(β -D-ribofuranosyl)pyrrolo[2,3-d]pyrimidine (500 mg, 1.75 mmoles) was dissolved in 100 ml of methauolic NH_3 (saturated at $-10^\circ)$ and the solution was heated in a steel bomb at 150° for 3 hr. The solution was evaporated to dryness and the residue was recrystallized twice from water to yield 250 mg (54%) of analytically pure tubercidin as colorless crystals.

20C

Anal. Caled for $C_{11}H_{14}N_4O_4$: C, 49.7; H, 5.30; N, 21.0. Found: C, 49.8; H, 5.46; N, 20.9.

TABLE II Ultraviolet Absorption of Some 4-Substituted

 $7-(\beta-b-1)$ 1BOFURANOSTL) PYRROLO [2,3-d] PYRIMIOINES



Compil	R	λ_{\max}^{pUU} to μ	-	$\lambda_{\max}^{(a)UU}$, tit μ	ŀ					
V11	811	322	(22,100)	3(6)	(20, 700)					
		267	rti, 000)	230	(17, 800)					
V1a	Ci	273	(1,450)	270	(4,600)					
		22:1	(27, 400)	224	(22,600)					
VIIIio	$N(CH_{a})$	278	(15,400)	281	(17,700)					
		232	(17,800)							
$1 \mathrm{Mb}$	$SC11_2C_611_5$	3111)	±10,300a	296	(13.800)					
		264	(8,900)	247						
		222	(22, 800)							
X1	$\rm NHCH_{\odot}$	272	:13,600)	273	:13 (fiQQ)					
		228	(18,500)							
1Xa	SC11.	304	: (1,000)	2173	(13),700)					
		260	(12,000)	249	(5,030)					
V1114	11	2014	:17.800/	263	(18.100)					
		227	332,1009							
VIIIa	OCH_{2}	270	(6, 200)	263	:7,6001					
		223	(24.300)							
VIIIe	$\rm NC_5 H_{46}$	284	(17,000)	287	(18,900)					
		2017	:14, 10út							

This compound was found to possess the same ultraviolet, infrared, and pure spectra and identical R_4 values in two chromatographic solvent systems (B, 0.51, and C, 0.53) as naturally occurring tubercidin.²⁹ There was also no depression observed by mixture melting point.

4-Amino-3-methyl-7-(β -D-ribofuranosyl)pyrrolo[2,3-d]pyrimidine (X),—Tubercidin²⁹ (I, 5.0 g, 0.0188 nole) was added to 50 nul of dimethylacetamide and 5 nul of methyl iodide was udded to the mixture. The mixture was stirred at room temperature for 24 hr (after less than 1 hr all the solid had dissolved). The solution was poured into 800 nul of ethyl ether with vigorous stirring and an oil formed which solidified upon standing. The solid was crushed, filtered from the mixture, and dried to yield 7.6 g (100%) of the hydriodide salt. This product was dissolved in ethauol (1.0 g/50 ml) and the volume was reduced to two-fifths of the original volume on a hot plate. The product slowly crystallized to yield 5.6 g (74%) of pure compound: mp 203–205°; $|\alpha|^{250} = 41.5^{\circ}$ (c 1, H₂O); – ultraviolet, $\lambda_{max}^{oH1} = 272 \text{ m}\mu$ ($\epsilon 9900$) and 226 (36,300); $\lambda_{max}^{M04} = 227 \text{ m}\mu$ ($\epsilon 22,900$), 270 (12,300), and 205 (sb) (5300).

4-Methylamino-7-(β -D-ribofuranosyl)pyrrolo[2,3-d]pyrimidine (XI). Method A.—4-Chloro-7-(β -D-ribofuranosyl)pyrrolo[2,3-d]-pyrimidine (VIa, 500 mg, 1.76 mmoles) was dissolved in ethanol (25 ml) containing anhydrons methylamine (2 ml) and the solution refluxed for 1 hr. The solution was evaporated to dryness and the residue was recrystallized from water to yield 380 mg (7777) of analytically pure product, mp 173-175.5°, { α }²³⁰ - (69.9° (c 1, 1777 HCl). The product was dried at 100° (0.1 mm) over P₂O_b for analysis.

Method B.--4-Amino-3-methyl-7- $(\beta$ -p-ribofuranosyl)pyrrolo-[2,3-d]pyrimidine hydriodide (N, 5.0 g) was dissolved in 30 ml of 1 N Nat)H and the solution was heated on a steam bath for 1 hr. The solution was then filtered and the filtrate was neutralized with 6 N HCl. The product crystallized from solution after standing at 5° overlight and was removed by filtration. The colorless solid was washed with a small amount of cold water and dried. Recrystallization from water furnished 3.08 g of product, up 175-177°.

(29) Purchased from the Uppola Co., Kalamazoo, Miele

The products obtained from methods A and B possessed identical ultraviolet, infrared, and pmr spectra and the same R_{γ} values in three chromatographic solvent systems (A, 0.66); B, 0.77; D, 0.80. A mixture melting point showed an depression:

4-Methoxy-7-(β -o-ribofuranosyl)pyrrolo[2, [-d]pyrimidine (VIIIa). -- 4-Uhloro-7-(β -o-rihofaranosyl)pyrrolo(2, [3-d]pyrimidine (VIIa). -- 4-Uhloro-7-(β -o-rihofaranosyl)pyrrolo(2, [3-d]pyrimididine (VIa, 4.0)g, 0.014 mole) was dissolved in 70 ml of 0.5 M methanolic NaOCH₂ and the solution refluxed for 1 hr. The solution was then cooled and metralized with 6 N/HCL. The actual unixture was evaporated to a thick shury *in caran* over a hotwater bath and the shury was dried by repeated evaporation *in caran* with 2-propanol. The dry residue was extracted with five 25-ml portions of acetome. The acetome extracts were concentrated and the product crystallized. A yield of 3.0 g ($76C_1^{\circ}$) of analytically pure product was obtained. A small sample was recrystallized from acetome, mp 158/460°, $\{\alpha\}^{2}(e-69.8)^{\circ}$ ir 1, Π_2 0.

 $A_{10}ab_{c}$ Caled (or $Y_{12}\Pi_{15}N_{h}O_{h}$; C, 51.3; H, 5.38; N, 15.0, Found: C, 51.5; H, 5.40; N, 15.2.

4-Dimethylamino-7-(β -o-ribofuranosyl)**pyrrolo**[2,3-d]**pyrini**dine (**VIIIb**), 4-Chloro-7-(β -o-ribofuranosyl)**pyrrolo**]2,3-d[**pyri**midine (**VIIa**, 2.9 g, 0.007 mode) was dissolved in methanol (80 micontaining anhydroas dimethylamine (5 ml) and the resolution solution was allowed to stand at room temperature for 1 hr. The solution was then evaporated *in room* temperature for 1 hr. The solution was then evaporated *in room* to yield an off which was dissolved in to ml of ethanol and again evaporated *in cacuo*. This procedure was repeated until the oil solidified and all trace of dimethylamine was removed. The crude product was dissolved in 50 ml of holing anhydrous ethanol and the solution was concentrated to 30 ml. The solid was removed by fibration, to yield 1.7 g (81) and ytically pure product, mp 102–1957, $|w|^{24} = 67.6^{2+} c$, 1420 s.

Auad. Caled for C₆₃H₀N₄O₅: C. 55.1; H. 6.17, N. 093. Found: C. 55.3; H. 6.60; N. 19.5.

4-(1-**Piperidy**])-**7**-(β -n-**ribofuranosy**])**pyrrolo**[2,3-d]**pyrimidine** (**VIIIc**) ψ 4-Chlor ϕ -7-(β -n-ribofuranosyl)**pyrrolo**[2,3-d]**pyrimidine** (VIa, 5.0 g, 0.0175 mole) was dissolved in 100 ml of ethanol containing 3.5 ml (2.92 g, 0.035 mole) of piperidine. The solution was reflexed for 1 he and then evaporated *in rareat* to yield an oil. The oil was thereaghly triturated with ethyl ether until the product solidined. The solid was completely pulverized in ethyl ether and after being removed by filtration was washed thoroughly with ethyl ether. The dried, colorless product was recrystallized twice from water to yield 3.3 g (55%) of colorless, analytically pure needles, mp 168 (70)⁵, { α }²⁵n + 73.5⁵ w 4, 0.1 N HCD.

7- $(\beta$ -D-**Ribofuranosyl)pyrrolo[2,3**-d]**pyrimidine** (VIIId, **7**-Deazanebularine), --4-Chloro-7- $(\beta$ -n-ribofuranosyl)pyrrolo[2,3-d]pyrimidine (2.5 g, 8.76 mmoles) was dissolved in a solution of 100 ml of ethanol and 5 ml of *curcearrated* NH₄OH. To this solution was added 1.0 g of π_{c} Pd C. The mixture was hydrogenated at 2.8 kg/cm² (40 psi) at room temperature for 4 hr and fibered through a Celite pail and the carbon was washed with five 5-ml portions of hot ethanol. The fibrate was taken to dryness *in varian*. The residue was dissolved in accone and carefully precipitated with petrolemm ether (hp 90-110⁶) to yield 1.5 g of a white solid (68⁶), mp 117°. A sample was dried at 100² over P₂O₈ (0.1 mart for anal ysis.

Anal. Caled for C₄₄H₄₈N₃O₅: C, 52,59; H, 5,22; N, 16,72;
 Found: C, 52,58; H, 5,48; N, 16,57.

7-(β -0-Ribofuranosyl)pyrrolo[2,3-d]pyrimidine-4-thiol (VII).

4-Chlora-7-(β -o-ribofuratiosyl)µyrrola[2,3-d]pyrimidine (VIa, 2.5 g) was dissolved in 39 ml of warm water containing 2.0 g of thiomrea. One drop of 25% acqueous farmic acid was added to the solution and the solution was heated for 15 min on a steam bath. After the first 5 min of heating the pH of the solution was adjusted to approximately 3 with concentrated NH₄011. The solution was then allowed to cool and the product crystallized. A yield of 1.7 g of crystals, up 206–208°, was obtained. An additional 400 mg of product was obtained by partial evaporation of the filtrate. The combined products were recrystallized from water to yield 1.7 g (68%) of analytically pure needles, up 206–208°, [n]²²0 –117.5° (r 0.65, 0.1 N NaO11), lit. ¹⁶ mp 204–207°, Anal. Cadd for C₀(Π_{12} NaOaS), C, 46.7; H, 4.63; N, 14.9. Found: C, 46.5; H, 4.44; N, 14.7.

 $\begin{array}{l} \textbf{4-Benzylthio-7-}(\beta\text{-}\upsilon\text{-}ribofuranosyl)pyrrolo[2,3-d]pyrimidine \\ \textbf{(IXb),}\text{-}\text{-}7-}(\beta\text{-}\upsilon\text{-}Ribofuranosyl)pyrrolo[2,3-d]pyrimidine \\ \textbf{4-thiol} \end{array}$

(VII, 500 mg, 1.77 mmoles) was dissolved in 4 ml of 14% aqueons NH₃. Benzyl chloride (230 mg, 1.8 mmoles) in 2.0 ml of dioxane was added to the basic solution and the mixture was stirred at room temperature for 3 hr. The mixture was diluted with 6 ml of water and then extracted five times with 7.5-ml portions of ethyl acetate. The ethyl acetate solution, after drying (MgSO₄), was evaporated *in vacuo* to yield an oil which solidified when triturated with ligroin (bp 9D–120°). The crude solid was recrystallized from an acetone–ligroin mixture to yield 500 mg (76° () of analytically pure product, mp 143–144°, $[\alpha]^{22}D = 66.3^{\circ}$ to 1, ethanol).

. (nul. Calcd for $C_{18}H_{19}N_3O_4S$: C_i 57.9; H, 5.13; N, 11.3. Fo.md: C, 58.1; H, 5.03; N, 11.1.

4-Methylthio-7-(β -D-ribofuranosyl)pyrrolo[2,3-d]pyrimidine (IXa).—To a solution of 0.55 g of NaOCH₃ in 30 ml of methanol was added 2.8 g of 7-(β -D-ribofuranosyl)pyrrolo[2,3-d]pyrimidine-4-thiol (VII). After VII had dissolved, 0.6 ml of methyl iodide in 5 ml of methanol was added and the solution was stirred at room temperature for 2 hr. The pH was then adjusted to 6 and the solution (some solid present) evaporated to dryness *in vacuo* at room temperature. The resulting yellow oil was tritunated with acetone. The white solid was collected by filtration, dried, and recrystallized from methanol to yield 2 g (65%) of analytically pure compound, mp 193–194°.

Anal. Calcd for $C_{12}H_{15}N_3O_4S$; C, 48.4; H, 5.05; N, 14.14. Found: C, 48.53; H, 5.20; N, 14.27.

Nucleosides. V. 2-Thiopyrimidine β -D-Arabinofuranosides

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The reaction of 2,2'-anhydro-1-(5-o-trityl- β -D-arabinoforanosyl)uracil (IIa) with H₂S in alkaline medium followed by detritylation produced 1-(β -D-arabinofuranosyl)-2-thionracil (IX). Thiation of the triacetate of IV yielded the corresponding 2,4-dithionracil nucleoside VI. Reaction of VI with NH₃ yielded the 2-thiocytosine arabinoside VII, cytosine arabinoside VIII, or the 2,2'-anhydrocytosine nucleoside IX depending on the conditions used. Reaction of IV or VII with bromine water resulted in the formation of 2,2'-anhydronucleosides IIb and IX, respectively. Iodination of IV gave a similar result. 1-(β -D-Arabinofuranosyl)-2-thiocytosine (VII) showed antiviral activity against vaccinia in tissue culture. The other thiopyrimidine nucleosides were inactive as antiviral agents.

Interest in thiopyrimidine nucleosides was heightened recently by the isolation of 4-thiouridylic acid¹ and a 2thiopyrimidine nucleotide^{2,3} from E. coli t-RNA as "odd" nucleotides and by the demonstration of an enzymatic thiolation of t-RNA.⁴ It has been suggested that² the facile formation and cleavage of a disulfide bond in t-RNA may provide a chemical mechanism to the "adapter modification hypothesis,"⁵ although no difference in the maximum tyrosine-accepting ability was detectable between the native and disulfide forms of E. coli tyrosine t-RNA.⁶ More recently, evidence was presented for reversible conformational changes and ribosome binding efficiency upon iodine oxidation of lysyl t-RNA from B. subtilis.⁷ A model involving sulfhydryl-disulfide interconversion of thiopyrimidines was again postulated. In our study of nucleoside antimetabolites, it was noted that the free base, 2thiouracil, has been reported to suppress the production of infective turnip yellow mosaic viral nucleoprotein,⁸ possibly via a preferential inhibition of viral-RNAdependent RNA synthesis.⁹ It was also incorporated into RNA of tobacco leaves and tobacco mosaic virus. The physicochemical difference between 2-thiouracil and uracil was indicated by recent nmr studies which concluded that 2-thiouracil exists essentially in the thiol form.¹⁰ The ultraviolet absorption spectrum of

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2-thiouridine suggests that it may act as an analog of cytidine as well as of uridine in RNA synthesis.⁹ Following our previous investigations of pyrimidine nucleosides as potential antiviral agents the synthesis of 2-thiopyrimidine β -D-arabinosides was considered of some interest.

In contrast to the 4-thiopyrimidine nucleosides which are readily available from the corresponding 4-oxopyrimidine nucleosides by thiation with $P_2S_{5,1}^{11,12}$ the 2thiopyrimidine nucleosides have been prepared by more circuitous routes.¹³ Methods involving the use of glycosyl animes^{14,15} and glycosylthioureas¹⁶ as starting points for building the 2-thiouracil ring system have been reported.

The formation of 2-thiouridine by the reaction of H_2S with 2,5'-anhydrouridine under mildly alkaline conditions has been described by Todd¹⁷ and by others.^{18,19} The bacterial synthesis of thiouridylic acid has been demonstrated.²⁰

Taking advantage of our recent experience with 2_{2}^{2} -

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